

IDENTIFICATION OF DERIVATIVES OF IONIZATION CONSTANTS 5-METHYLTHIENO[2,3-*D*]PYRIMIDINE-4(3*H*)-OH-6-CARBOXILIC ACID

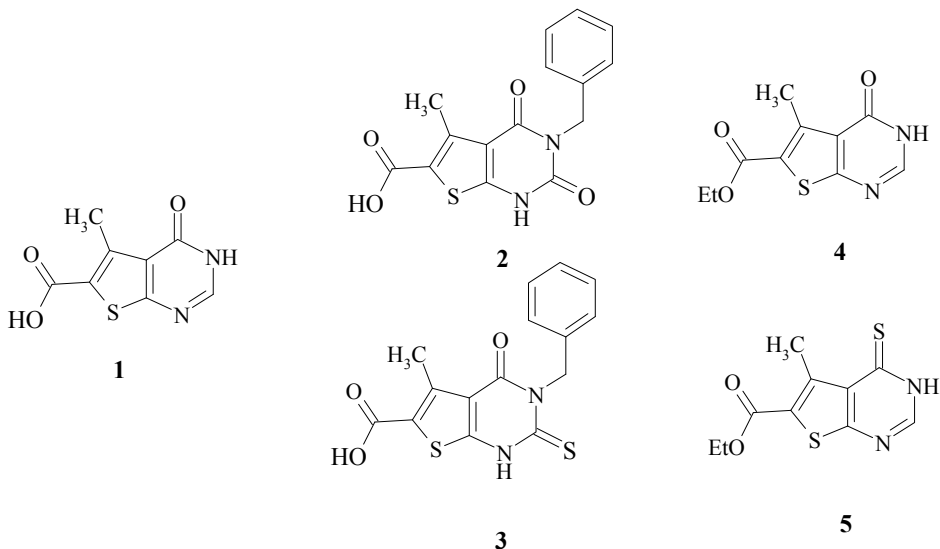
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Derivatives of 5-methylthieno[2,3-*d*]pyrimidine-4(3*H*)-on-6-carboxylic acid are biologically active compounds. However, these compounds in the un-ionized form are soluble in water, making it difficult to introduce them into dosage forms. For further selection of salt formation cation we have determined the ionization constants of compounds **1-5** by alkalimetryc non-aqueous titration. The media for determining was the mixture of dioxane-water (1:1), equivalence point had been determined potentiometrically. Results were processed statistically.



As a result, it was found that acids **1-3** are strong enough dibasic acids. We found, that acid **3** is the strongest acid, which contains a Sulfur atom in position 2.

Ionization constants for the 5-methylthieno-[2,3-*d*]pyrimidine-4(3*H*)-(thi)on-6-carboxylic acids esters **4** and **5** have been determined additionally, it was shown that compound **5** reveals more acidic properties.

Thus it was found that the most suitable for salt formation with the basics, including weak ones are compounds containing sulfur atom in the pyrimidine nucleus.